

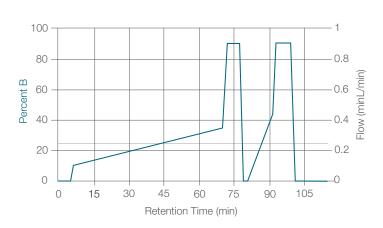
APPLICATION NOTE

## Mobile phase solvents and reagents for LC-MS biopharmaceutical applications

Biopharmaceutical drugs are a unique class of therapeutic proteins that have far more complex structures than small molecule drugs. Liquid chromatography (LC) coupled with mass spectrometry (MS), LC-MS, has become increasingly popular for characterizing biopharmaceutical drugs. LC-MS is widely used for characterizing the primary structure of protein drugs such as amino acid sequencing for peptides/proteins, profiling of process-related impurities such as host cell proteins (HCPs), disulfide bond mapping, N-glycan profiling of monoclonal antibodies (mAbs), and nucleotide sequencing for oligonucleotides.

At the primary structure level, the protein post-translational modifications (PTMs) that arise at different stages of manufacturing and storage can impact drug efficacy and safety; thus, PTMs are classified as critical quality attributes (CQA) that should be monitored and controlled. Combining the power of LC separation with the sensitivity and specificity of MS detection, LC-MS is the tool of choice for peptide mapping to detect multiple PTMs in a single analytical run, i.e., the multi-attribute method (MAM) workflow [1].

A critical component of the MAM workflow is to separate peptides of various lengths from the digested therapeutic protein. This is typically achieved by employing a long LC gradient of water and acetonitrile with formic acid added as a modifier, as shown in Figure 1. Mobile phase solvents and reagents must be of high purity with low organic impurities to ensure no interference with detection by electrospray (ESI) MS.



Time	% В
0	1
5	1
6	10
70	35
72	90
77	90
79	1
81	1
83.5	10
91.5	45
93	90
99	0-
101	1
115	1

Figure 1. Typical LC gradient in a MAM workflow for the separation of NIST mAb peptides with a binary mobile phase: A: 0.1% formic acid in water (Cat. No. LS118); B: 0.1% formic acid in acetonitrile (Cat. No. LS120), from Reference [1].



Thermo Fisher Scientific offers two grades of LC-MS solvents. Fisher Chemical™ Optima™ grade LC-MS solvents are high purity and meet all the needs for LC-MS analysis. Pre-made LC-MS blends are also available in Fisher Chemical Optima grade and are precisely mixed with lot-to-lot consistency and tested by LC-MS, making these blends convenient for use in biopharma QC labs. The Thermo Scientific™ UHPLC-MS grade solvents are a higher grade with more stringent specifications to meet the most sensitivity-demanding analyses. Both LC-MS and UHPLC-MS grades of solvents and blends have been fully tested to ensure low organic impurities to minimize interference and ion suppression.

Another important quality attribute of our high-purity LC-MS regents is the extremely low levels of metal impurities. Trace metal impurities can cause poor chromatography peak shape, peak tailing, and diminished recovery via the metal-ion mediated adsorption on the LC column media [2]. The alkali metal ions such as sodium (Na+) and potassium (K+) are electrostatically attracted to the negatively charged carboxyl group of the C-terminal amino acid in peptides/ proteins and the polyanionic backbone of oligonucleotides [3]. The Na+ and K+ adducts decrease the overall detection sensitivity and complicate the peptide identification in LC-MS analysis. Minimizing the metal contamination in mobile phase solvents and regents has become ever more critical with the growing demand for higher LC-MS sensitivity.

All Thermo Fisher Scientific LC-MS grade solvents and reagents are subject to strict control of trace metal contamination during the production process and thoroughly tested to stringent specifications with ICP-MS. A comparison of the four most common alkali and alkaline contaminant metal impurities in pre-blended LC-MS grade 0.1% formic acid in water from three competitors are listed along with the Fisher Chemical Optima offering in Table 1. As shown, the Fisher Chemical Optima pre-blended mobile phase has the lowest allowable overall metal ions present.

Table 1. Specification of alkali and alkaline metals in four LC-MS grade 0.1% formic acid in water (in ppb) from four vendors.

Metal impurity	Fisher Chemcial Optima (Cat. No. LS118)	Competitor M	Competitor H	Competitor J
Na	≤50	≤700	≤5000	≤50
K	≤ 50	≤40	≤2000	≤50
Mg	≤10	≤40	≤500	≤50
Ca	≤50	≤50	≤500	≤50

The Fisher Chemical Optima LC-MS grade mobile phase additives are equally thoroughly tested and subject to the same strict quality control paradigm as the solvents. As shown in Table 2, the Optima LC-MS grade trifluoroacetic acid (Cat. No. A116) also has the lowest overall allowable metal ion content among the six different LC-MS grade trifluoroacetic acids surveyed. Note that two competitor offerings do not even have specifications on metal ions.

Table 2. Metal impurity specifications of six different LC-MS grade trifluoroacetic acid offerings (in ppb) from five vendors.

Metal impurity	Fisher Chemical Optima (Cat. No. LS118)	Competitor M	Competitor H	Competitor H (Ultra)	Competitor R	Competitor P
Na	≤50	≤500	≤500	≤500	No spec	No spec
K	≤20	≤100	≤100	≤100	No spec	No spec
Mg	≤10	≤500	≤500	≤500	No spec	No spec
Ca	≤50	≤200	≤200	≤200	No spec	No spec

## Purity grades for every LC-MS application

Fisher Chemical Optima LC-MS Grade Solvents: general purpose for all LC-MS applications, low organic and metal impurities, tested with stringent LC-MS gradient suitability tests.

Description	Packaging	Quantity	Cat. No.	
Acetonitrile	Amber Glass Bottles	500 mL, 1 L, 2.5 L, 4 L	A955	
Methanol	Amber Glass Bottles	500 mL, 1 L, 2.5 L, 4 L	A456	
Water	Amber Glass Bottles	500 mL, 1 L, 2.5 L, 4 L	W6	
2-Propanol	Amber Glass Bottles	500 mL, 1 L, 2.5 L, 4 L	A461	

Thermo Scientific™ UHPLC-MS Grade Solvents: ultrapure quality solvents to meet your most demanding UHPLC and LC-MS applications for low background and high sensitivity.

Description	Packaging	Quantity	Cat. No.
Acetonitrile	Clear Borosilicate Glass Bottles	1 L	A956-1
Methanol	Clear Borosilicate Glass Bottles	1 L	A458-1
Water	Clear Borosilicate Glass Bottles	1 L	W8-1

Thermo Scientific UHPLC-MS Reagent Installation Kit (Cat. No. UHPLCMSKIT): recommended for setting up and passing the specification tests during the installation of a new LC-MS system. The kit includes:

- 1 L Acetonitrile (Cat. No. A956-1)
- 2 × 1 L Methanol (Cat. No. A458-1)
- 2 × 1 L Water (Cat. No. W8-1)
- 1 L Thermo Scientific<sup>™</sup> ChromaCare<sup>™</sup> Instrument Flush Solution (Cat. No. T111101000): a blend of 25% acetonitrile, 25% methanol, 25%, water, and 25% 2-propanol (IPA) to clean the flow path of an LC-MS system.



Fisher Chemical Optima LC-MS Grade Mobile Phase Blends: precisely pre-blended for convenience and tested for lot-to-lot consistency.

Description	Packaging	Quantity	Cat. No.
Acetonitrile with 0.1% FA	Amber Glass Bottle	500 mL, 1 L, 2.5 L, 4 L	LS120
Acetonitrile with 0.1% TFA	Amber Glass Bottle	500 mL, 1 L, 2.5 L, 4 L	LS121
Water with 0.1% FA	Amber Glass Bottle	500 mL, 1 L, 2.5 L, 4 L	LS118
Water with 0.1% TFA	Amber Glass Bottle	500 mL, 1 L, 2.5 L, 4 L	LS119

Fisher Chemical Optima LC-MS Grade Mobile Phase Additives: tested for high impurity and low trace metals for LC-MS use.

Description	Packaging	Quantity	Cat. No.
Formic Acid	Poly Bottles	50 mL	A117-50
	Ampules	0.5, 1, 2, 10 × 1 mL	A117
Acetic Acid	Poly Bottles	50 mL	A113-50
	Ampules	0.5, 1, 2, 10 × 1 mL	A113
Trifluoroacetic Acid	Amber Glass Bottle	50 mL	A116-50
	Ampules	0.5, 1, 2, 10 × 1 mL	A116
Ammonium Formate	Glass Bottle	50 g	A11550
Ammonium Acetate	Glass Bottle	50 g	A11450

## References

- 1. Haichuan Liu, et al, "A high-resolution accurate mass multi-attribute method for critical quality attribute monitoring and new peak detection," Thermo Scientific Application Note 72916 (2019)
- 2. Robert E. Birdsall, et al, "Application of mobile phase additives to reduce metal-ion mediated adsorption of non-phosphorylated peptides in RPLC/MS based assays," Journal of Chromatography B, Vol. 1126-1127 (2019)
- 3. Joachim Emmert, et al, "Influence of Na\* and K\* Concentration in Solvents on Mass Spectra of Peptides in LC–ESI-MS," Spectroscopy, Vol 21, Issue 2, Feb 1 (2006)

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